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G-000-104 .119

**SUBMITTAL OF FMPC PROJECT 6-053, SCRAP PICKLING
FACILITY FOR A DETERMINATION UNDER 40 CFR 61.06 AS TO
WHETHER EPA ADMINISTRATOR APPROVAL IS REQUIRED FOR
PROJECTED WORK**

12/15/88

DOE-303-89
DOE-FN USEPA
22
LETTER



Department of Energy

Oak Ridge Operations
P.O. Box 2001
Oak Ridge, Tennessee 37831—

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December 15, 1988
DOE-303-89

Mr. David Kee, Director
Air and Radiation Division
U.S. Environmental Protection Agency
Region V, Mail Code 5AC-26
230 South Dearborn Street, 13th Floor
Chicago, IL 60604

Dear Mr. Kee:

SUBMITTAL OF FMPC PROJECT 6-053, SCRAP PICKLING FACILITY FOR A DETERMINATION UNDER 40 CFR 61.06 AS TO WHETHER EPA ADMINISTRATOR APPROVAL IS REQUIRED FOR PROJECTED WORK

Reference is made to a letter dated January 11, 1988, from J. A. Reafsnyder to Bill Franz, subject: "Demonstration of Compliance with National Emission Standards for Hazardous Air Pollutants (NESHAPS) - Feed Materials Production Center (FMPC)."

Enclosed is an application for a determination by the administrator under 40 CFR 41.06 as to whether the equipment to be installed is a modification of the source (the entire FMPC site) as defined by 40 CFR 61.15. If so, please consider the submittal as an application to modify under the requirements of 40 CFR 61.07. This is a very minor source of the same magnitude as the fourteen (14) submitted on August 9, 1988, and the two submitted on November 8, 1988, for a similar determination.

The application format followed is that required by the Ohio EPA for permits to install (PTIs). Calculation of dose equivalents follows the same methodology as that in the previous submittals and is based on the 1986 computations provided in the computer printouts enclosed with the reference letter.

We would appreciate a timely review of this application under 40 CFR 61.06 and 61.07 if the latter is applicable. If you have any questions or require additional information, please contact Weldon Dillow of our Environmental Protection Division at FTS 626-1354.

Sincerely,

Ray Hansen
James A. Reafsnyder
Site Manager

SE-311:Dillow

See Page 2 for Enclosure and ccs.

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Mr. David Kee

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Enclosure:

FMPC No. 6-053, Scrap Pickling Facility--

Application for a Determination if Installation
of FMPC No. 6-053 is a modification pursuant to
40 CFR 61.15.

cc w/o enclosure:

L. C. Goidell, SRO

M. L. Galper, WMCO

R. F. Hodanbosi, OEPA

J. C. Tseng, EH-23, FORSTL

M. Neal, DP-84

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ENCLOSURE

FMPC Source 6-053

SCRAP PICKLING FACILITY

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ENCLOSURE

METHODOLOGY FOR CALCULATING THE DOSE TO MAXIMALLY EXPOSED MEMBERS OF THE PUBLIC FROM PLANT 6 SCRAP PICKLING FACILITY (6-053)

No separate calculations for Plant 6 were made for the 1986 emissions, therefore, the emissions and committed dose equivalents for the entire FMPC site for 1986 are used in calculating the committed dose equivalents for Plant 6 equipment. The uranium site emissions for 1986 were as follows:

Uranium Emissions

U-234 - 9.10×10^{-3} Ci or 1.47 g
U-235 - 4.53×10^{-4} Ci or 212 g
U-238 - 9.72×10^{-3} Ci or 29200 g

Total U emissions - 1.93×10^{-2} Ci or 29412 g

Committed dose equivalents from the FMPC 1986 emissions which include uranium, fission products, radium, thorium, neptunium, and plutonium are:

Effective committed dose equivalent -- 1.18 mrem
Lung committed dose equivalent -- 7.38 mrem
Endosteal bone committed dose equivalent -- 7.14 mrem

Then the committed dose equivalents from the Plan 6 Scrap Pickling Facility which are estimated to result in a net increase of 0.3 pounds or 136 grams of uranium plus accompanying radionuclides per year, are:

Effective Committed Dose Equivalent:

$$\frac{1.36 \times 10^2 \text{ g} \times 1.18 \text{ mrem}}{2.941 \times 10^4 \text{ g}} = 5.5 \times 10^{-3} \text{ mrem/yr}$$

Lung Committed Dose Equivalent:

$$\frac{1.36 \times 10^2 \text{ g} \times 7.38 \text{ mrem}}{2.941 \times 10^4 \text{ g}} = 3.41 \times 10^{-2} \text{ mrem/yr}$$

Endosteal Bone Committed Dose Equivalent:

$$\frac{1.36 \times 10^2 \text{ g} \times 7.14 \text{ mrem}}{2.941 \times 10^4 \text{ g}} = 3.30 \times 10^{-2} \text{ mrem/yr}$$

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Although these are not rigorous calculations, they are accurate enough to indicate that this facility will have a very minor effect on site emissions and will not affect compliance with the site dose equivalent standards. The numbers calculated above compare favorably with those using 1987 data. For example, in 1987 35.4 Kg of uranium plus accompanying radionuclides were emitted with a calculated effective committed dose equivalent of 1.2 mrem. Then the effective committed dose equivalent for the Plant 6 Scrap Pickling Facility is

$$\frac{1.36 \times 10^2 \text{ g from Plant 6} \times 1.2 \text{ mrem}}{3.54 \times 10^4 \text{ g from entire plant}} = 4.6 \times 10^{-3} \text{ mrem}$$

which is comparable to the 5.5×10^{-3} mrem using 1986 data.

Check Appropriate Box(es)

- Air Discharge
- Water Discharge to New Source Treatment Works
- Solid Waste Disposal Facility
- Hazardous Waste Disposal Facility

For Office Use Only

PTI Application No. _____

Date Received _____

Premise No. _____

OHIO ENVIRONMENTAL PROTECTION AGENCY
Application for Permit to Install

US Department of Energy Feed Materials Production Center

Applicant's Name

PO Box 398705

Mailing Address

Cincinnati, Hamilton Ohio 45239

City County State Zip Code Telephone Number

Mary Stone, DOE/FMPC (513) 738-6656

Person to contact (Name and Title and Telephone Number)

FMPC is located near Fernald, Ohio at 7400 Willey Road

Location of Proposed Facility (State the location as completely and precisely as possible)*

FMPC Plant 6

Crosby Township Hamilton County 45030

City or Township County Zip Code

Directions: A Permit to Install is required for new or modified sources of pollution under the provisions of OAC Rule 3745-31. An application cannot be considered complete unless all applicable questions are answered and the required information has been submitted. This application must be signed in accordance with OAC Rule 3745-31-04(B) or it cannot be accepted.

Applicants for permits involving air emissions or wastewater treatment facilities will be required to pay a permit to install fee as shown in Section 3745.11(B) and (C) of the Ohio Revised Code. This fee is payable fifteen days after the date of final issuance of the permit.

Name of new or modified source or facility: Ingot and Scrap Pickling Facility Plant (6-053)

Product of new or modified source/facility: Surface cleaned Uranium Ingots and Scrap

Will the proposed source/facility involve any of the following: Check all that apply.

- A. Air Discharge
- B. Wastewater Treatment Works
- C. Solid Waste Disposal Facility
- D. Hazardous Waste Disposal Facility

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*Example: "The source will be constructed on a 20 Acre plot to be located on Franklin Township Road No. 17, approximately 1 1/4 miles north of the intersection of State Route 99 and Franklin Township Road No. 17"

Under OAC 3745-31-04, These signatures shall constitute personal affirmation that all statements or assertions of fact made in the application are true and complete, comply fully with applicable state requirements, and shall subject the signatory to liability under applicable state laws forbidding false or misleading statements.

Ray Hansen 12/15/88
Authorized Signature (for facility) Date
James A Reafsnyder
Site Manager

Title
U. S. Department of Energy
P.O. Box 398705
Cincinnati, OH 45239-8705
Address

For Wastewater
Treatment Plants:

Signature of General Contractor or Agent Date
Performing installation, if selected.

Company

Address

The following information must be submitted on a separate piece of paper and attached to this application.

FOR ALL PERMITS TO INSTALL

1. Describe the product or service to be produced by the applicant along with a description of the proposed source/facility.
2. List the name and quantity of all materials and chemicals (solid, liquid, or gaseous) that will be used or produced by the source/facility.
3. State the reason for the application. Is this a new installation, modification to an existing source/facility, reconstruction of an existing source/facility, or startup of a source/facility that has been permanently shutdown for ____ year (State number of years)
4. Has a previous Ohio EPA application or plan submission been filed for this source facility? If so, state the date and type of the application previously submitted
5. Will the proposed source/facility comply with all rules, laws, and regulations of Ohio EPA and U.S. EPA?

FOR AIR POLLUTION SOURCES

6. State the amount of each air contaminant (actual emissions) from each source in pounds per hour and tons per year at maximum and average conditions.
7. Are the proposed sources required to comply with the following federal requirements:
 - i. New Source Performance Standards (NSPS)
 - ii. National Emission Standards for Hazardous Air Pollutants (NESHAPS)
 - iii. Prevention of Significant Deterioration (PSD)
 - iv. Appendix "S" - Emission Offset Policy
8. Will the proposed sources employ best available technology?
9. Will the proposed sources cause the significant degradation of air quality?
10. Will the proposed sources interfere with the attainment and maintenance of the ambient air quality standards?
11. Describe any source monitoring, emission monitoring, or control equipment monitoring devices to be installed by the applicant.
12. Will the proposed sources involve the use of asbestos, benzene, beryllium, mercury, or vinyl chloride?
13. Complete and attach an anticipated construction schedule for each proposed source (See attached).
14. Please include the estimated cost of any air pollution control equipment to be installed on the proposed sources.
15. An appendix for each air contaminant source must accompany this application. See the following description of the appendices, determine which should accompany your application.

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1. Describe the product or service to be produced by the applicant along with a description of the proposed source/facility.

The Feed Materials Production Center (FMPC) is a government owned facility which produces highly pure uranium metal. The proposed air emission source to be installed at the FMPC Plant 6 is an ingot and scrap pickling process. Plant 6 is a production facility used for heat treating, acid pickling, and finish machining of uranium metal.

The new Plant 6 ingot and scrap pickling process utilizes 8 normal (N) nitric acid to surface clean cast ingots or scrap uranium metal. The pickling process removes decay deposits and oxidation from the ingots and scrap metal. The cleaned ingots are machined while the scrap metal is used for recasting.

2. List the name and quantity of all materials and chemicals (solid, liquid, or gaseous) that will be used or produced by the source/facility.

The ingot and scrap pickling process utilizes a 8 N nitric acid solution to surface clean the uranium metal. The facility uses a 500 gallon nitric acid tank and 2 water rinse tanks which are replenished on an as needed basis.

3. State the reason for the application. Is this a new installation, modification to an existing source/facility, reconstruction of an existing source/facility, or startup of a source/facility that has been permanently shutdown for _____ years?

The ingot and scrap pickling process is a new installation at the FMPC.

4. Has a previous Ohio EPA application or plan submission been filed for this source, facility? If so, state the date and type of the application previously submitted.

An OEPA permit to install (PTI) application for this air emission source was submitted to SWOAPCA on October 1, 1987. On January 5, 1988, SWOAPCA was provided with a revised Appendix A-2 for the ingot and scrap pickling process. The January 1988 submittal reflected the final design specifications on the NOx Destructor which vented the scrap pickling process. On July 20, 1988, SWOAPCA requested potential radionuclide emissions for this source. Additional production, emission, and engineering information was obtained in an effort to respond to this request. This revised application reflects the most current system design and includes an Appendix A which lists uranium as a pollutant controlled by the Destructor.

5. Will the proposed source/facility comply with all rules, laws, and regulations of Ohio EPA and U.S. EPA?

Yes.

FOR AIR POLLUTION SOURCES

6. State the amount of each air contaminant (actual emissions) from each source in pounds per hour and tons per year at maximum and average conditions.

See attachment for emission calculations. A summary table is shown below.

Summary Table

| | <u>NO_x Emissions</u> | <u>Uranium Emissions</u> |
|--------------------------------------|-------------------------------------|--------------------------|
| A. Average Controlled (lbs/hr) | 18.2 (NO ₂) 2.0 (NO) | 2.14 X 10 ⁻⁵ |
| B. Maximum Controlled (lbs/hr) | 18.2 (NO ₂) 2.0 (NO) | 3.23 X 10 ⁻⁵ |
| C. Average Controlled (tons/yr) | 18.2 (NO ₂) 2.0 (NO) | 9.00 X 10 ⁻⁵ |
| D. Maximum Controlled (tons/year) | 54.6 (NO ₂) 6.0 (NO) | 1.36 X 10 ⁻⁴ |

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FOR AIR POLLUTION SOURCES (CONT.)

7. Are the proposed sources required to comply with the following federal requirements?

- i. - No
- ii. - Yes. Based on updated production, emissions, and engineering information, the new scrap pickling process is estimated to release at maximum operating conditions approximately 0.3 pound of uranium per year.
- iii. - No
- iv. - No

8. Will the proposed sources employ best available technology?

The proposed source will employ best available technology for NO_x and uranium removal. The pickling system will be enclosed, hooded, and vented to a NO_x Destructor (packed tower design). The design efficiency for uranium and NO_x removal is estimated to be 95%. In addition, the metal is vacuumed to remove loose surface contamination prior to pickling. The vacuum exhaust is vented to a HEPA filter.

9. Will the proposed sources^{cause} the significant degradation of air quality.

The source does not cause a significant degradation of air quality.

10. Will the proposed sources interfere with the attainment and maintenance of the ambient air quality standards?

The source does not interfere with the attainment and maintenance of the ambient air quality standard.

11. Describe any source monitoring, emission monitoring, or control equipment monitoring devices to be installed by the applicant.

The ingot and scrap pickling process will be ducted to a NO_x Destructor (packed tower design). A FUJI infrared monitor (model 1155) will be used to measure NO_x emissions from the Destructor.

12. Will the proposed sources involve the use of asbestos, benzene, beryllium, mercury, or vinyl chloride?

No.

13. Complete and attach an anticipated construction schedule for each proposed source (See Attached).

Attached.

14. Please include the estimated cost of any air pollution control equipment to be installed on the proposed sources.

Estimated cost of the NO_x Destructor is \$400000.

15. An appendix for each air contaminant source must accompany this application. From the following description of the appendices, determine which should accompany your application.

Attached.

OHIO ENVIRONMENTAL PROTECTION AGENCY

INSTALLATION SCHEDULE

TO ACCOMPANY APPLICATION FOR PERMIT TO INSTALL

THIS FORM CONSTITUTES PART OF THE APPLICATION OF:

FACILITY NAME: Feed Materials Production Center

ADDRESS: 7400 Willey Road, Fernald, OH 45030

FOR A PERMIT TO INSTALL THE FOLLOWING AIR CONTAMINANT SOURCE:

IDENTIFICATION: Plant 6 INGOT AND SCRAP PICKLING FACILITY (6-053)

DESCRIPTION: Surface cleaning of cast Ingots and scrap utilizing nitric acid

THE INSTALLATION OF THE ABOVE AIR CONTAMINANT SOURCE IS PLANNED TO FOLLOW THE TIME SCHEDULE DESCRIBED BELOW:

| | <u>DATE</u> |
|------------------------------------|-------------|
| 1. EQUIPMENT ORDERED - - - - - | <u>4/88</u> |
| 2. COMMENCE CONSTRUCTION - - - - - | <u>1/89</u> |
| 3. STARTUP - - - - - | <u>6/89</u> |
| 4. PERFORMANCE TESTING - - - - - | <u>6/89</u> |

Premise No.
 Source No.
 Application No.

APPENDIX A, PROCESS

PROCESS DATA

1. Name of process Ingot and Scrap Pickling Facility - 6 (6-053)
2. End product of this process Pickled (Surface Cleaned) Uranium Ingot & Scrap
3. Primary process equipment Acid Pickling Tanks and Rinse Tanks
 Your identification Ingot and Scrap Pickling Facility - Plt. 6 Year Installed 1988
4. Manufacturer Custom Design Make or model NA
5. Capacity of equipment (lbs./hr): Rated 8730 Max. 13,200
6. Method of exhaust ventilation: Stack Window fan Roof vent
 Other, describe _____
 Are there multiple exhausts? Yes No

OPERATING DATA

7. Normal operating schedule: 24 hrs./day, 7 days/wk., 50 wks./year.
8. Percent annual production (finished units) by season:
 Winter 25 Spring 25 Summer 25 Fall 25
9. Hourly production rates (lbs.): Average 8730 Maximum 13,200
10. Annual production (indicate units) 55,440 tons
 Projected percent annual increase in production 10%
11. Type of operation: Continuous Batch
12. If batch, indicate Minutes per cycle 15 Minutes between cycles 5
13. Materials used in process:

| List of Raw Materials | Principal Use | Amount (lbs./hr.) |
|-----------------------|--|-------------------|
| 8N Nitric Acid | Pickling (Surface Cleaned) Uranium Ingots and Scrap | 14.7 lbs/hr |
| | | |
| | | |
| | | |
| | | |

14. A PROCESS FLOW DIAGRAM MUST BE INCLUDED WITH THIS APPENDIX. Show entry and exit points of all raw materials, intermediate products, by-products and finished products, Label all materials including airborne contaminants and other waste materials. Label the process equipment and control equipment.

CONTROL EQUIPMENT

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Control Equipment Code:

- | | | |
|--------------------------------|--------------------------|---------------------------|
| (A) Settling chamber | (G) Cyclonic scrubber | (M) Adsorber |
| (B) Cyclone | (H) Impingement scrubber | (N) Condenser |
| (C) Multiple cyclone | (I) Orifice scrubber | (O) Afterburner - catal |
| (D) Electrostatic precipitator | (J) Venturi scrubber | (P) Afterburner - them |
| (E) Fabric filter | (K) Plate or tray tower | (Q) Other, describe _____ |
| (F) Spray chamber | (L) Packed tower | |

15. Control Equipment data:

| Item | Primary Collector | Secondary Collector |
|---|---|---------------------|
| (a) Type (See above code) | L1 & L2 (2 in series) | |
| (b) Manufacturer | Tri-Mer Corp. | |
| (c) Model No. | N/A | |
| (d) Year installed | 1988 | |
| (e) Your identification | N-06H-3000, 4000 | |
| (f) Pollutant Controlled | NOx, Uranium | |
| (g) Controlled pollutant emission rate (if known) | NOx, Uranium See Attached Calculations | |
| (h) Pressure drop | N/A | |
| (i) Design efficiency | NO2-97% NO1-95% Uranium-95% | |
| (j) Operating efficiency | NOx-95% Uranium-95% | |

STACK DATA

16. Your stack identification EP6-11
17. Are other sources vented to this stack? Yes No
If yes, identify sources Scrap Pickling, Chip Pickling, and Ingot Pickling
18. Type: Round, top inside diameter dimension 36"
 Rectangular, top inside dimensions (L) _____ x (W) _____
19. Height: Above roof 15 ft., above ground 60 ft.
20. Exit gas: Temp. 68 °F, Volume 25,000 ACFM, Velocity 4,000 ft./min.
21. Continuous monitoring equipment: Yes No
If yes, indicate: Type NDIR Mass Flow Detector, Manufacturer Whittaker
Make or Model FUJI 760, Pollutant(s) monitored NO, NO2
22. Emission data: Emissions from this source have been determined and such data is included with this appendix: Yes No
If yes, check method: Stack Test Emission factor Material balan

Completed by J. S. Straub, Date 8/28/87
 1st Revision by T. J. Walsh 11/12/87
 2nd Revision by R. Cauley/T.J. Walsh/D.A. Nixon 10/14/88

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EMISSION DATA FOR APPENDIX A-2

QUESTION #22

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Emissions Calculations for Plant 6 Scrap Pickling

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1. The new scrap pickling facility will be operated in a similar manner as the existing operation.
2. The emissions tests conducted on the existing operation are applicable to the new installation. The calculations for the new Plant 6 Scrap Pickling process use an average loading factor of 4.61×10^{-4} mg/dscf for 3740 lbs/hr of uranium processed.
3. Engineering estimate of uranium removal efficiency for the source's control equipment is 95%.
4. Flow rate from the scrap pickling process to control equipment is 3000 dscf/min.

Calculations

1. Average loading factor (mg/dscf)

$$4.61 \times 10^{-4} \text{ mg/dscf} \times 8730 \text{ lbs of material /hr (new process) =}$$

$$\frac{\text{-----}}{3740 \text{ lbs of material /hr during testing}}$$

(existing)

$$1.08 \times 10^{-3} \text{ mg/dscf expected from new process}$$

2. Average hourly uncontrolled uranium emissions (lbs/hr)

$$1.08 \times 10^{-3} \text{ mg/dscf} \times 3000 \text{ dscf/min} \times 60 \text{ min/hr} \times .001 \text{ g/mg} \times 1 \text{ lb/454g} =$$
$$4.28 \times 10^{-4} \text{ lbs/hr}$$

3. Maximum loading factor (mg/dscf)

$$4.61 \times 10^{-4} \text{ mg/dscf} \times 13200 \text{ lbs of material processed/hr} =$$

$$\frac{\text{-----}}{3740 \text{ lbs of material processed/hr during testing}}$$

$$1.63 \times 10^{-3} \text{ mg/dscf}$$

4. Maximum hourly uncontrolled uranium emissions (lbs/hr)

$$1.63 \times 10^{-3} \text{ mg/dscf} \times 3000 \text{ dscf/min} \times 60 \text{ min/hr} \times .001 \text{ g/mg} \times 1 \text{ lb/454g} =$$
$$6.46 \times 10^{-4} \text{ lbs/hr}$$

5. Average yearly uncontrolled uranium emissions (tons/yr)

$$4.28 \times 10^{-4} \text{ lbs/hr} \times 24 \text{ hrs/day} \times 7 \text{ days/wk} \times 50 \text{ wks/yr} \times 1 \text{ ton/2000lbs} =$$
$$1.80 \times 10^{-3} \text{ tons/yr}$$

6. Maximum yearly uncontrolled uranium emissions (tons/yr)

$$6.46 \times 10^{-4} \text{ lbs/hr} \times 24 \text{ hrs/day} \times 7 \text{ days/wk} \times 50 \text{ wks/yr} \times 1 \text{ ton/2000lbs} =$$
$$2.71 \times 10^{-3} \text{ tons/yr}$$

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7. Average hourly controlled uranium emissions (lbs/hr)

4.28×10^{-4} lbs/hr \times (1-.95) removal efficiency = 2.14×10^{-5} lbs/hr

8. Maximum hourly controlled uranium emissions (lbs/hr)

6.46×10^{-4} lbs/hr \times (1-.95) removal efficiency = 3.23×10^{-5} lbs/hr

9. Average yearly controlled uranium emissions (tons/yr)

1.80×10^{-3} tons/yr \times (1-.95) removal efficiency = 9.00×10^{-5} tons/yr

10. Maximum yearly controlled uranium emissions (tons/yr)

2.71×10^{-3} tons/yr \times (1-.95) removal efficiency = 1.36×10^{-4} tons/yr

A. Hourly NO_x Emission

Estimate of hourly NO_x emission was based on input data provided by the FMPC's Operating Contractor (NO_x-rich air stream Input rate is 25,000 SCFM at 3,960 ppm NO_x). It is further assumed that the maximum hourly emission rate equals the average hourly emission rate.

Calculate NO₂ Emission Rate

Based on vendor's performance limit of 100 ppm NO₂ in the clean exhaust air and use of the ideal gas law, the hourly NO₂ emission rate is calculated as follows: (SCFM measured at 1.0 atm and 60°F).

$$(1) \rho = \frac{P}{RT} \quad \text{where } \rho = \text{density, } P = \text{pressure, } R = \text{specific gas constant and } T = \text{temperature (K)}$$

$$(2) R = \frac{R_u}{M} \quad \text{where } R_u = \text{universal gas constant and } M = \text{molar mass}$$

$$\text{From equation (2): } R = \frac{R_u}{M} = \frac{8314 \text{ J/kg-mol} \cdot \text{K}}{46 \text{ g/g-mol}} = 181 \text{ J/kg} \cdot \text{K}$$

Calculate density of NO₂ from equation (1):
Assume cubic feet measured at 1.0 atm and 60°F

$$\rho = \frac{P}{RT} = \frac{1.0 \text{ atm}}{(181 \text{ J/kg} \cdot \text{K})(60^\circ\text{F})}$$

$$\text{where } P = 1.0 \text{ atm} = 1.0133 \times 10^5 \text{ N/m}^2$$

$$R = 181 \text{ N} \cdot \text{m} / \text{kg} \cdot \text{K}$$

$$T = ^\circ\text{K where, } T = 273^\circ\text{K} + t (^{\circ}\text{C})$$

$$\text{Convert } 60^\circ\text{F to } ^\circ\text{C from } t(^{\circ}\text{C}) = (60^\circ\text{F} - 32^\circ\text{F}) / 1.8 \frac{^\circ\text{F}}{^\circ\text{C}}$$

$$\rho = \frac{(1.0133 \times 10^5 \text{ N/m}^2)}{(181 \text{ N} \cdot \text{m/kg} \cdot \text{K})(289\text{K})} = 1.94 \text{ kg/m}^3$$

To calculate hourly emission rate:

$$\left(\frac{100 \text{ ft}^3 \text{ NO}_2}{10^6 \text{ ft}^3 \text{ of air}} \right) \times \left(\frac{25000 \text{ ft}^3 \text{ of air}}{\text{min}} \right) \times \left(\frac{60 \text{ min}}{\text{hr}} \right) \times \left(\frac{1.94 \text{ kg}}{\text{m}^3} \right) \times$$

$$\left(0.06243 \frac{\text{m}^3}{\text{kg}} / \frac{\text{ft}^3}{\text{lb m}} \right) = 18.2 \text{ lb/hr of NO}_2$$

Calculate NO emission rate:

NO_x Conversion Equation =



$$3960 \text{ ppm}_v \text{ NO}_x \text{ (6 moles NO/72 mole NO}_x\text{)} = 330 \text{ ppm}_v \text{ NO}$$

From equation (2):

$$R = \frac{R_u}{M} = \frac{8314 \text{ J/kg} \cdot \text{mol} \cdot \text{K}}{30 \text{ g/g} \cdot \text{mol}} = 277 \text{ J/kg} \cdot \text{K}$$

Calculate density of NO from equation (1):
Assume cubic feet measured at 1.0 atm and 60°F

$$\rho = \frac{P}{RT} = \frac{1.0 \text{ atm}}{(277 \text{ J/kg} \cdot \text{K})(60^\circ\text{F})}$$

where

$$P = 1.0 \text{ atm} = 1.0133 \times 10^5 \text{ N/m}^2$$

$$R = 277 \text{ N} \cdot \text{m/kg} \cdot \text{K}$$

$$T = ^\circ\text{K where, } T = 273^\circ\text{K} + t \text{ (}^\circ\text{C)}$$

$$\text{Convert } 60^\circ\text{F to } ^\circ\text{C from } t(^{\circ}\text{C}) = (60^\circ\text{F} - 32^\circ\text{F}) / 1.8 \frac{^\circ\text{F}}{^\circ\text{C}}$$

$$\rho = \frac{(1.0133 \times 10^5 \text{ N/m}^2)}{(277 \text{ N} \cdot \text{m/kg} \cdot \text{K})(289\text{K})} = 1.27 \text{ kg/m}^3$$

To calculate hourly emission rate: (assume 95% removal efficiency)

$$\left(\frac{330 \text{ ft}^3 \text{ NO}}{10^6 \text{ ft}^3 \text{ of air}} \right) \times (1.00 - 0.95) \times \left(\frac{25000 \text{ ft}^3 \text{ of air}}{\text{min}} \right) \times \left(\frac{60 \text{ min}}{\text{hr}} \right) \times \left(\frac{1.27 \text{ kg}}{\text{m}^3} \right) \times \left(\frac{0.06243 \text{ m}^3}{\text{kg}} / \frac{\text{ft}^3}{\text{lb}_m} \right) = 2 \text{ lb/hr of NO}$$

B. Annual NO_x Emission from Scrubber

Average annual emission rate is calculated based on the assumption that the system will be operated 1 shift per day. Maximum annual emission occurs under the operating condition of 3 shifts per day.

Average annual NO₂ emission rate:

$$\text{NO}_2 = (18.2 \text{ lb/hr})(8 \text{ hr/shift})(1 \text{ shift/day})(5 \text{ days/wk})(50 \text{ wk/yr}) = \frac{36400 \text{ lb/yr}}{2000 \text{ lbs/ton}} = 18.2 \text{ ton/yr}$$

Average annual NO emission rate:

$$\text{NO} = (2 \text{ lb/hr})(8 \text{ hr/shift})(1 \text{ shift/day})(5 \text{ days/wk})(50 \text{ wk/yr}) = \frac{4000 \text{ lbs/yr}}{2000 \text{ lbs/ton}} = 2 \text{ tons/yr}$$

Maximum annual NO₂ emission rate:

6204

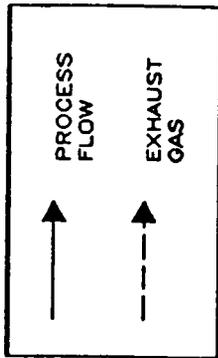
NO₂ = (18.2 lb/hr)(8 hr/shift)(3 shifts/day)(5 days/wk)(50 wk/yr) =

$$\frac{109200 \text{ lbs/yr}}{2000 \text{ lbs/ton}} = 54.6 \text{ tons/yr}$$

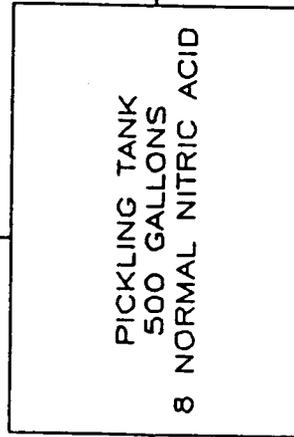
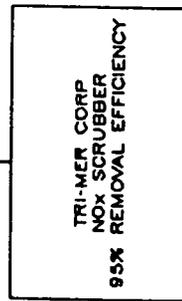
Maximum annual NO emission rate:

NO = (2 lb/hr)(8 hr/shift)(3 shifts/day)(5 days/wk)(50 wk/yr) =

$$\frac{12000 \text{ lbs/yr}}{2000 \text{ lbs/ton}} = 6 \text{ tons/yr}$$



25,000 CFM EXHAUST EMISSIONS - SEE ATTACHED CALCULATIONS



SURFACE CLEANED SCRAP URANIUM METAL
 TO PLANT 5 CASTING FURNACES
 SURFACE CLEANED URANIUM INGOTS
 TO PLANT 6 MACHINING

SCRAP URANIUM
 METAL AND INGOTS
 13,200 LB/HR MAX.

INGOT AND SCRAP PICKLING FACILITY - PLANT 6
 FEED MATERIALS PRODUCTION CENTER